

#### THE UNIVERSITY of TEXAS

SCHOOL OF HEALTH INFORMATION SCIENCES AT HOUSTON

## X-Ray Crystallography Pt. I

For students of HI 6001-125 "Computational Structural Biology"

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http://biomachina.org/courses/structures/02.html



- Incident and scattered plane waves
- Elastic scattering: No change in wavelength

Probing Macromolecular Atomic Structures

- Crystalline sample

- Probe should be chosen such that its wavelength is of atomic dimensions: X-rays

- Suitable detection systems

#### From Diffraction Spots to Atomic Models



#### Scattering from Regular Atomic Arrays

- Diffraction pattern may be understood from principles of optical diffraction on gratings

- Scattering from single atom
- Scattering from an atomic array

#### Path and Phase Difference



Path differencePhase difference $\mathbf{R} \cdot \mathbf{S}_{0} - \mathbf{R} \cdot \mathbf{S}_{I}$  $\frac{2\pi}{\lambda} \begin{bmatrix} \mathbf{R} \cdot \mathbf{S}_{0} - \mathbf{R} \cdot \mathbf{S}_{I} \end{bmatrix}$ 

#### Scattering from a Single Atom



#### Real and Reciprocal Space



real space/ direct space/ Cartesian space reciprocal space/ frequency space/ Fourier space

#### Phase Difference and Scattering Power



If electron at origin has a scattering power E(S) along direction S, Scattering power of an electron at R is 2π iS.R E(S) e Structure factor: e<sup>2π iS.R</sup> (Geometrical part of the scattering) SF for Electron Density Distribution

Structure factor for a unit volume:  $e^{2\pi i S.r}$ 

For a volume element drStructure factor:  $\rho(r)e^{2\pi i s.r} dr$ 

For a continuous electron density distribution,

Structure factor:  $F(S) = \int \rho(r) e^{2\pi i S.r} dr$  Plane Waves and Fourier Transforms

$$\mathbf{F(S)} = \int \rho(\mathbf{r}) e^{2\pi i \mathbf{S} \cdot \mathbf{r}} d\mathbf{r}$$

# **Structure factor is the Fourier Transform of the electron density distribution**

**Conversely, the electron density distribution is the inverse Fourier transform of the structure factor** 

$$\rho(\mathbf{r}) = \frac{1}{V} \int \mathbf{F}(\mathbf{S}) \mathbf{e}^{-2\pi \mathbf{i} \mathbf{S} \cdot \mathbf{r}} d\mathbf{S}$$

Phase Problem in Crystallography

The structure factor F(S) for a given direction S is a complex quantity.

 $\mathbf{F(S)} = |\mathbf{F(S)}| e^{i\phi}$ 

|F(S)| : Structure factor amplitudes↓ Phases

The measured intensity I(S) is I(S) = F(S) \* F(S) =  $|F(S)|^2$ 

Thus the phases \$\$ are lost during measurement.

Phases Needed to Reconstruct Images

#### **Scattered X-rays carry phase information**

But no X-ray lens to recombine the amplitudes and the phases to reconstruct the original image!

### Which is more important: Amplitudes or phases?

#### Phases more Important than Amplitudes



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#### Phase Problem in Crystallography

# How to estimate the phases that give the best electron density $\rho(r)$ from the structure factors F(S)

#### Uniform Sample Cannot Deflect X-rays

Atoms must have heterogenous electron density distribution to scatter X-rays.

Assume a uniform electron density distribution:

$$\rho(\mathbf{r}) = \rho$$
$$\mathbf{F(S)} = \int \rho(\mathbf{r}) e^{2\pi \mathbf{i} \mathbf{S} \cdot \mathbf{r}} d\mathbf{r}$$
$$= \rho \int e^{2\pi \mathbf{i} \mathbf{S} \cdot \mathbf{r}} d\mathbf{r}$$
$$= \rho \delta(\mathbf{S} - \mathbf{0})$$
$$= \mathbf{F(0)}$$

Uniform Sample Cannot Deflect X-rays

# F(S) = F(0) if $\rho(r) = \rho$

Thus, a uniform electron density gets scattered only in the forward direction.

Spatial or temporal heterogeneity that creates a contrast between the electrons and the background, is a must to scatter X-rays in all directions.

X-ray Scattering from a Single Atom

Assume a spherically symmetric electron density distribution



X-ray Scattering from a Single Atom

$$\mathbf{F(S)} = 2\pi \int_{0}^{\infty} dr \rho(r) r_{0}^{2} \int_{0}^{\pi} d\theta \sin\theta e^{2\pi i \operatorname{Sr} \cos\theta}$$

**Substitute**  $x = r\cos\theta$ 



**f(S):** Atomic form factor

Finite Atomic Size & Form Factor

Assuming a tailed electronic distribution  $\rho(r) = Zne^{kr^2}$  where Z is the atomic number it can be shown that  $f(S) = Ze^{\left(\frac{\pi^2}{k}\right)S^2}$ 

Thus, finite size of atom introduces a resolution dependent fall-off of the atomic form factor.

This is due to the destructive interference of the scattered radiation from electrons.

#### Resources

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Practical Guide. New York: Macmillan